

The background features abstract, colorful swirls in shades of purple, green, and blue, interspersed with small yellow triangles. The text is centered and rendered in a blue, serif font with a subtle drop shadow.

Beam Application to Nanotechnology based on Subpicosecond Pulse Radiolysis

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8-1 Mihogaoka, Ibaraki, Osaka 567-0047, Japan*

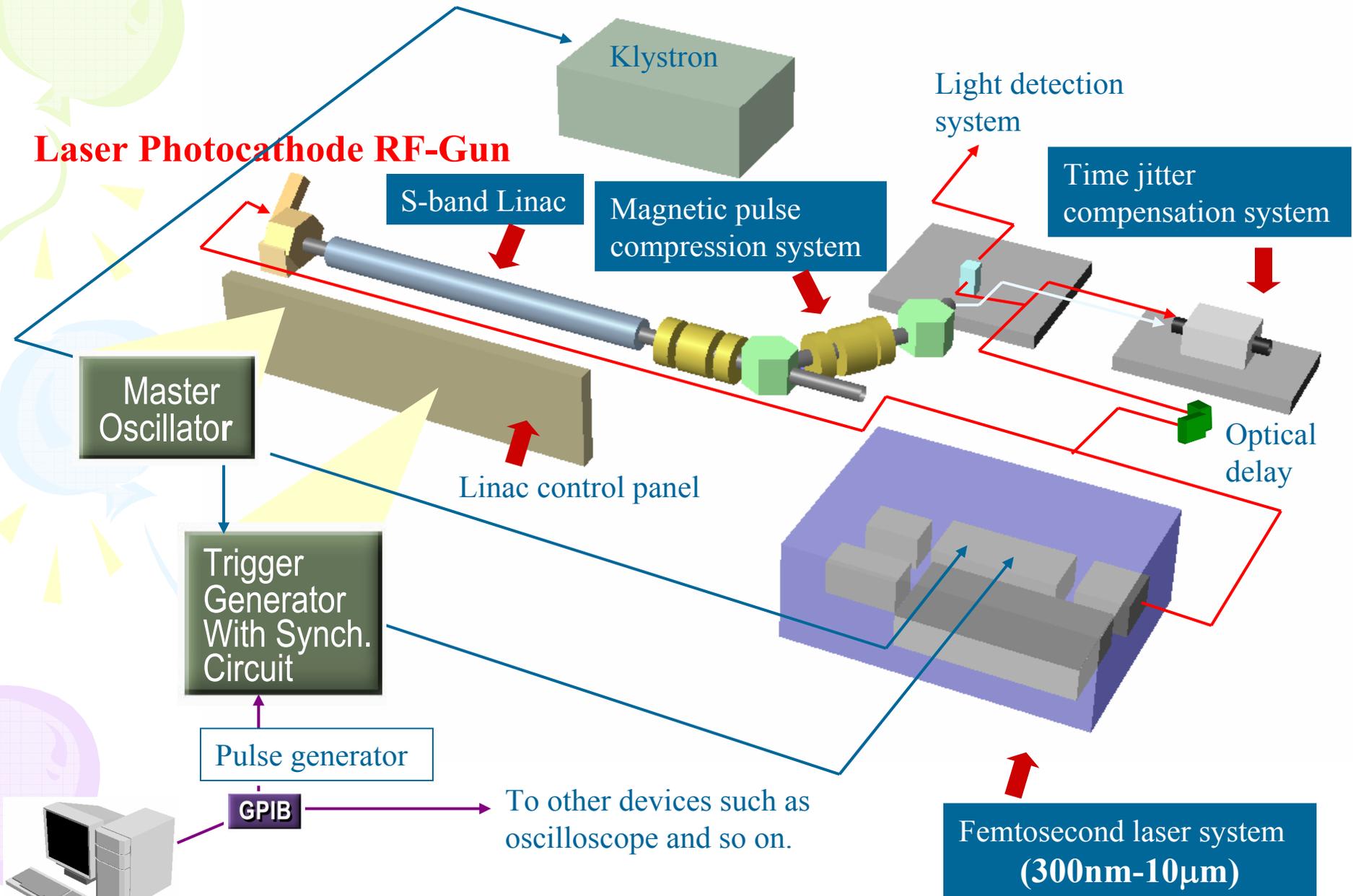


Acknowledgment

- **Coworkers**
 - T. Kozawa
 - A. Saeki
 - H. Yamamoto
 - K. Okamoto
 - Prof. S. Seki
 - Tsukuda
- 
- 

New Femtosecond Pulse Radiolysis System

Laser Photocathode RF-Gun



Co-operation of Our Research Groups in The Institute of Scientific and Industrial Research, Osaka University

**Department of Beam
Science & Technology**
(S. Tagawa, Y.Yamamoto,
K.Kobayashi, A.Saeki)

**Department of Beam
Processing for Nanotechnology**
(S.Tagawa, T.Kozawa, S.Seki)

Radiation Laboratory
20 Members
(S.Tagawa, Y.Yamamoto,
T.Kozawa, S.Seki,
K.Kobayashi, A.Saeki
from our group)

Nanofoundry
Y.Matsui and K.Okamoto
in Tight Co-operation with
S.Tagawa, T.Kozawa, S.Seki

21st Century COE Program
Chairman of 21st Century COE Program Committee (S. Tagawa)
5 Research Groups: IT Nanotechnology G (Group Leader : S.Tagawa)

Nanofabrication Research of Our Group

Based on **Radiation Chemistry**

**Top-down Type
Nanotechnology
Nanolithographies**

**Bottom-up Type
Nanotechnology
Material Design and synthesis**

**Beam Science
and Technology**

**Research
Consortiums**
SELETE(ASUKA) :
EB, F2, Immersion ArF
ASET : EUV
**Several
Companies**

**Combination of
Two Nanotechnologies
Mass-production Type
Nanofabrication**

**Next Generation Lithography
and Future Lithographies <30nm
(CD control <2nm)
(So-called Mass Production Type)**

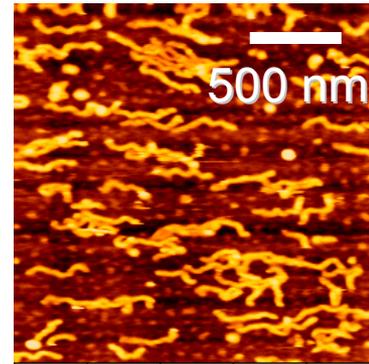
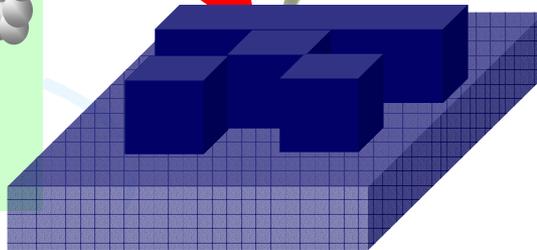
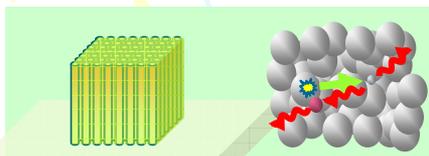
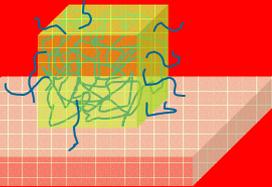
**Nanoscience and
Nanotechnology
(So-called Non-Mass Production Type)**

Nanobeam Processes and Development of Nanomaterials

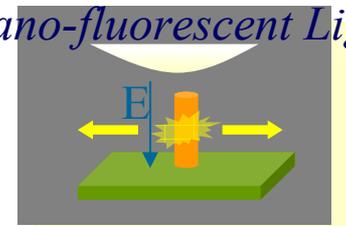
Ultra-fine Electron Beam pattern

Use of Single Charged Particle Events

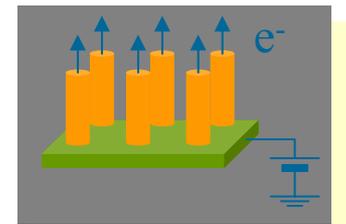
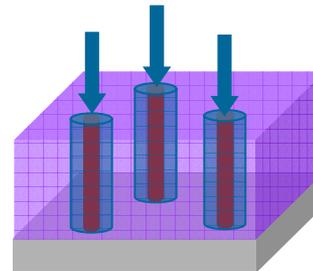
**Super Fine Pattern
Formation
by the design of chemical
reaction, molecular structure**



Nano-fluorescent Light



Single Ion Track



Field Emitter

Nano Materials

C60 etc.,
Carbon Nanotubes



1985
1991 Iijima

Size Control, Spatial Distribution, Structure, etc.

Radiation Induced Reactions in Polymers Induced Ions

The size of the field depends on
Reactive intermediates

Density

Stability

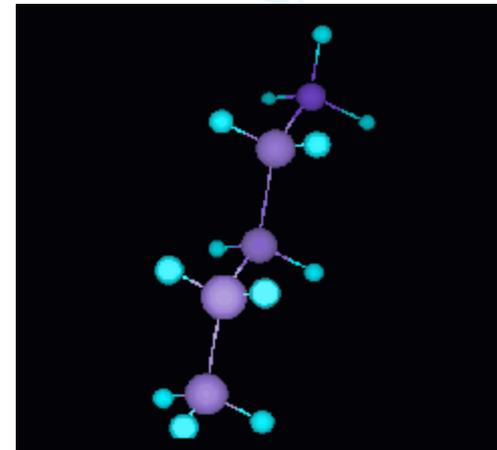
Linear energy transfer: LET

Structure of a molecule (Mw, Stiffness, etc.)

Very good size control and mass production

Ion Track Structure in Polysilanes

- Polysilanes
 - » Silicon Analog of Polyethylene
 - » 1-D Analog of Crystalline Si
- Properties
 - Electroluminescence
 - Photoconductivity
 - 1-D Quantum wires
 - Radiation Sensitive



Uniformity of Nano-Wires

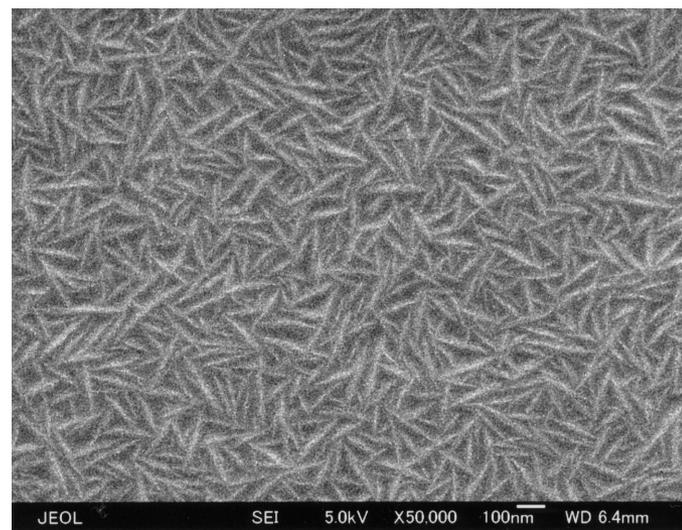
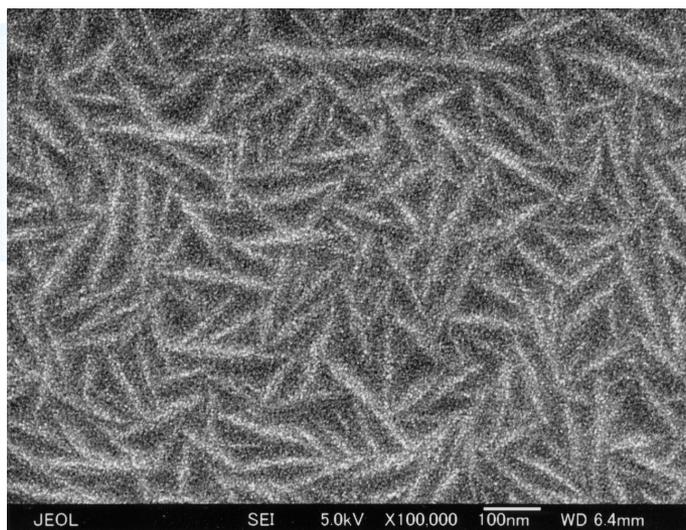


Figure. A SEM image of rod-like nano-wires on a Si substrate. The nano-wires were formed by the 450 MeV $^{128}\text{Xe}^{23+}$ irradiation to a **PS3** thin film (200 nm thick) at 1.6×10^{12} ions/cm². The film was developed by hexane, and heated up to 523 K for 0.5 h after irradiation.

Uniform Formation of Nanowires

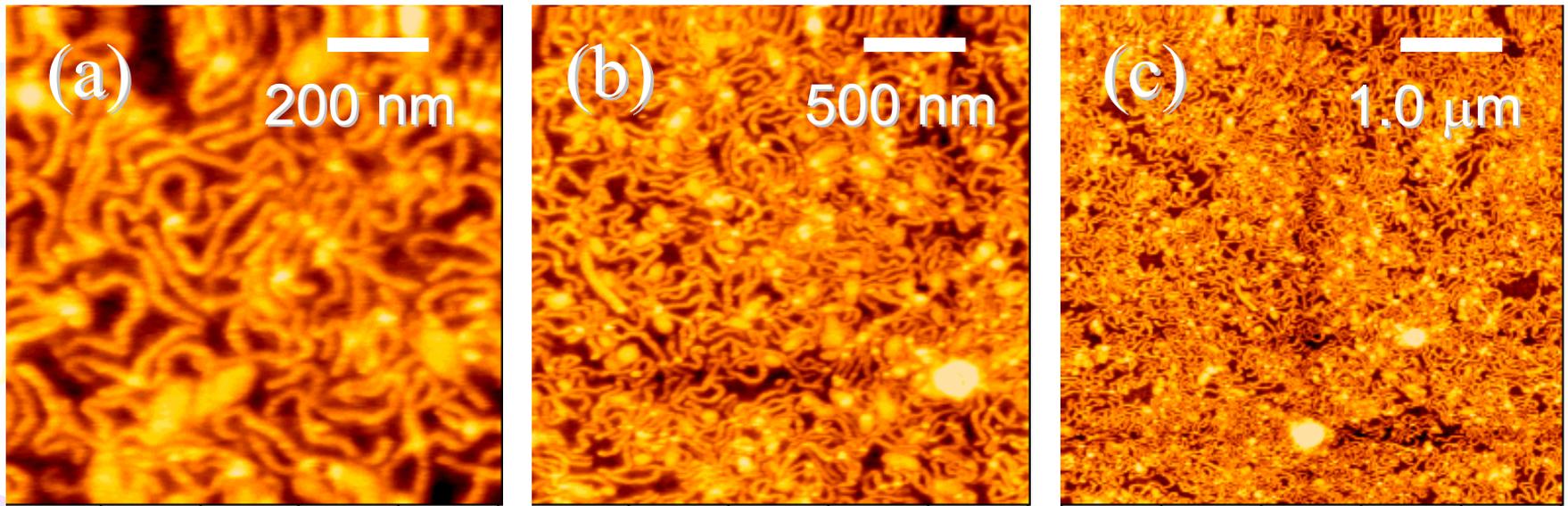
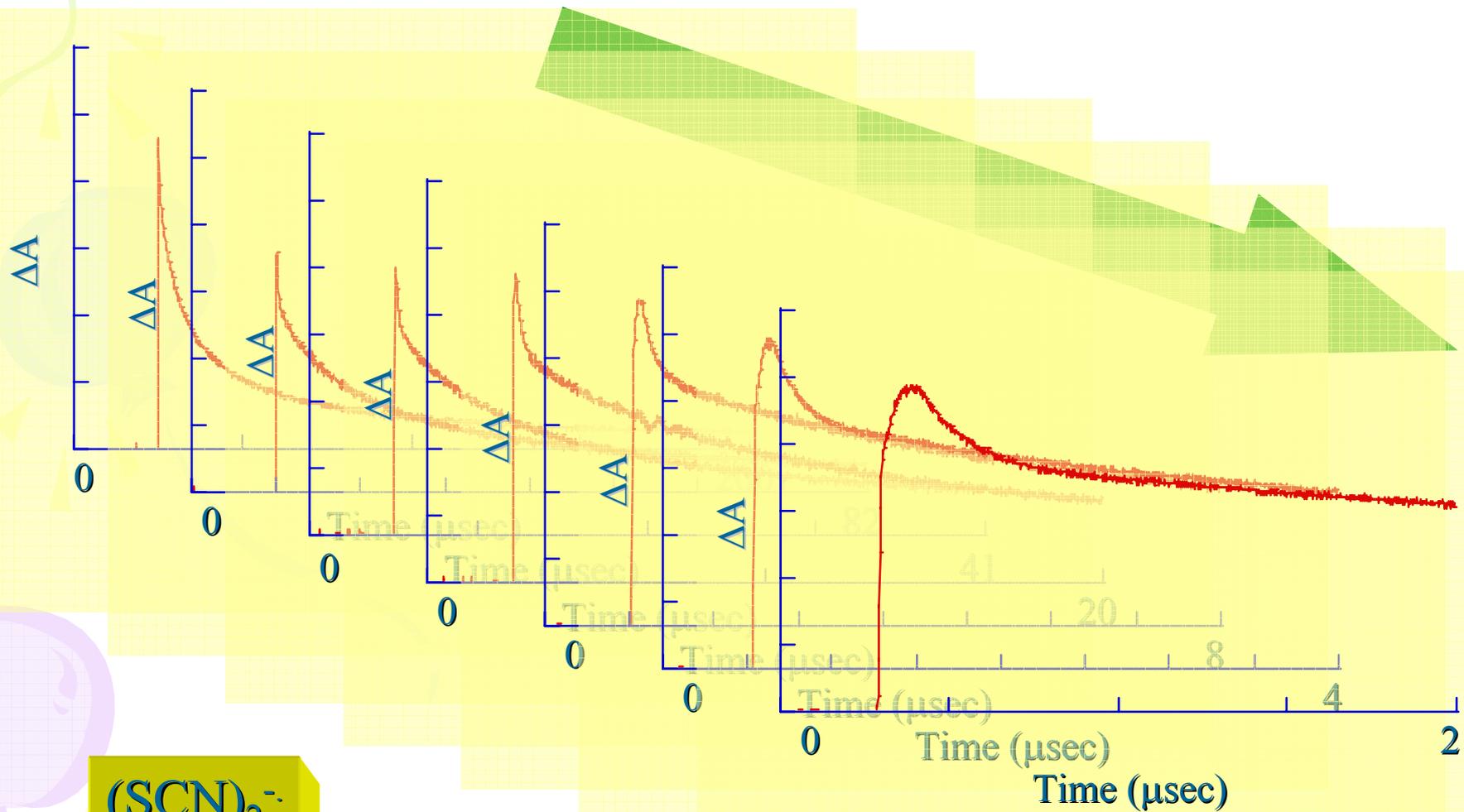
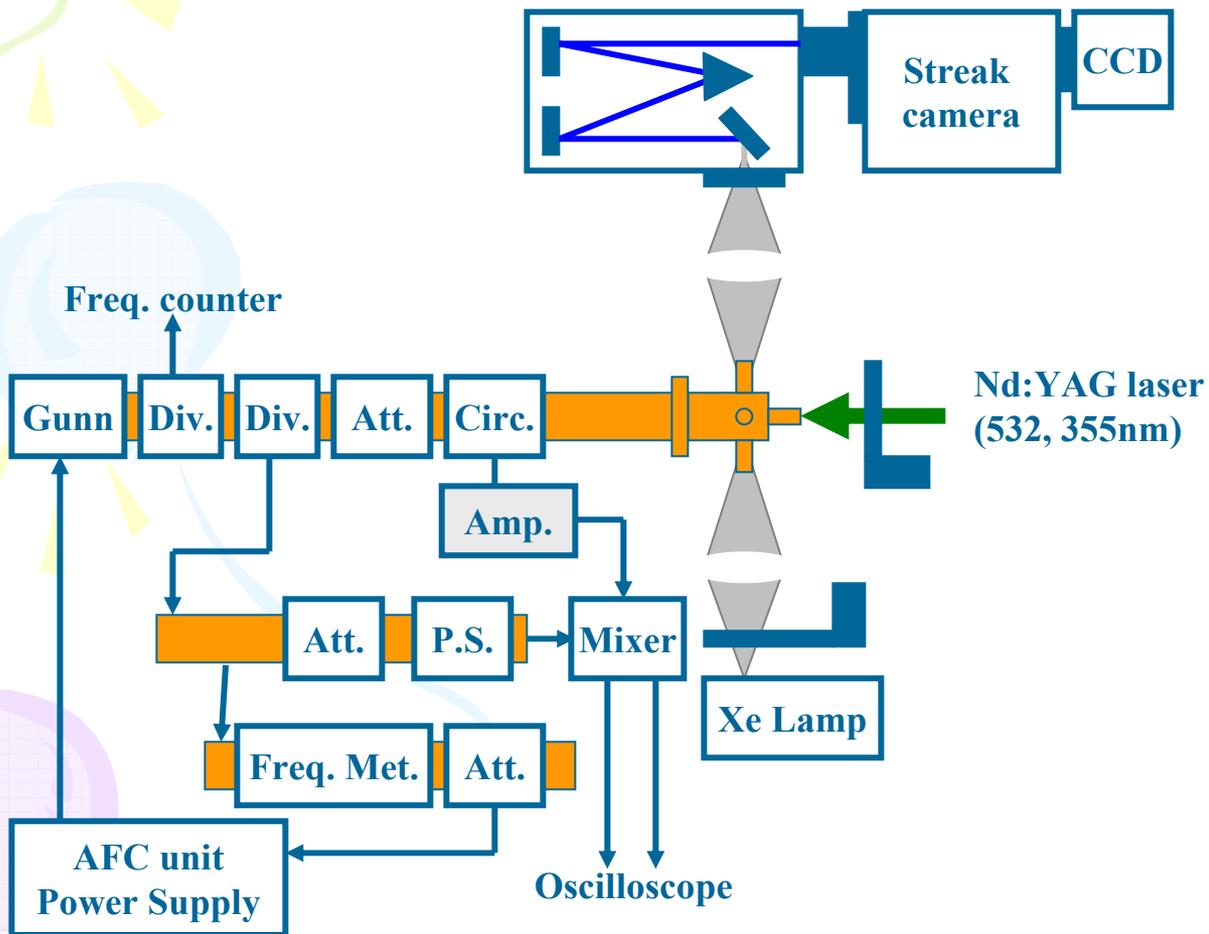


Figure. AFM images of nano-wires on a Si substrate with a variety of special resolutions. The nano-wires were formed by the 450 MeV $^{128}\text{Xe}^{23+}$ irradiation to a **PS1** thin film (0.40 μm thick) at 1.7×10^{10} ions/cm².

A kinetic trace in wide dynamic range can be measured by one pulse irradiation

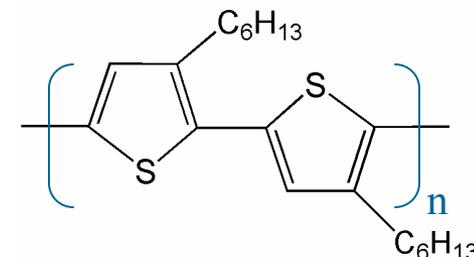


In-situ TRMC-TAS

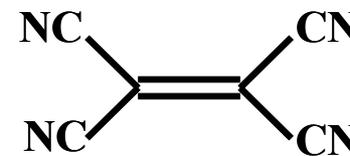


Materials (Film)

Dissolved in THF and drop-casted on quartz substrate.



RR-P3HT



TCNE

History of pulse radiolysis at ISIR

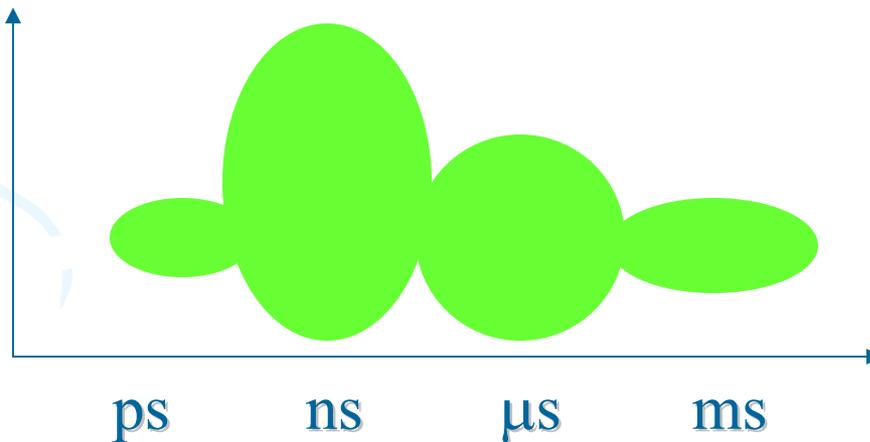
ns system

- UV, Vis 1994
- NIR, 1995
- Low-temp, NIR, Vis, 1997
- IR, 1998
- UV, Vis, ns-ms, 2001

ps system (stroboscopic)

- Laser-linac synchronized, 1995
- Vis, 1998
- Pulse compression, 1998
- Jitter compensation, 1999
- Improvement of S/N ratio, 2001

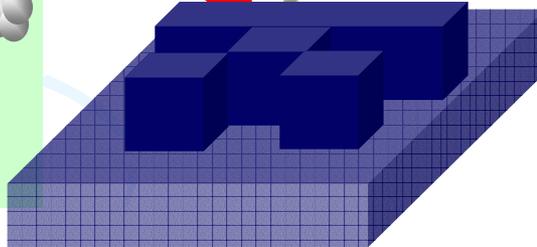
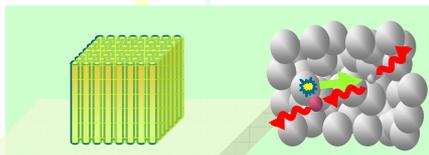
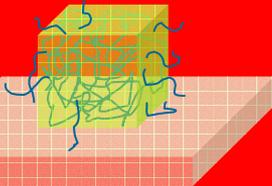
IR
NIR
Vis
UV



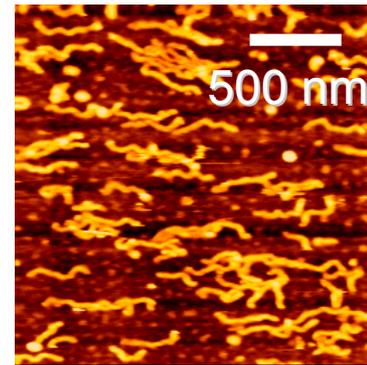
Nanobeam Processes and Development of Nanomaterials

Ultra-fine Electron Beams

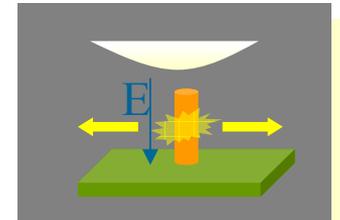
Super Fine Pattern Formation
by the design of chemical reaction, molecular structure



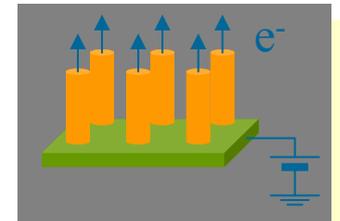
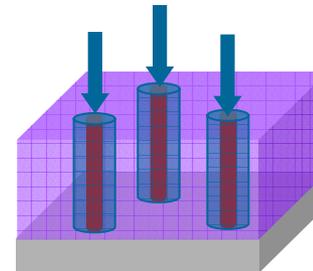
Use of Single Charged Particle Events



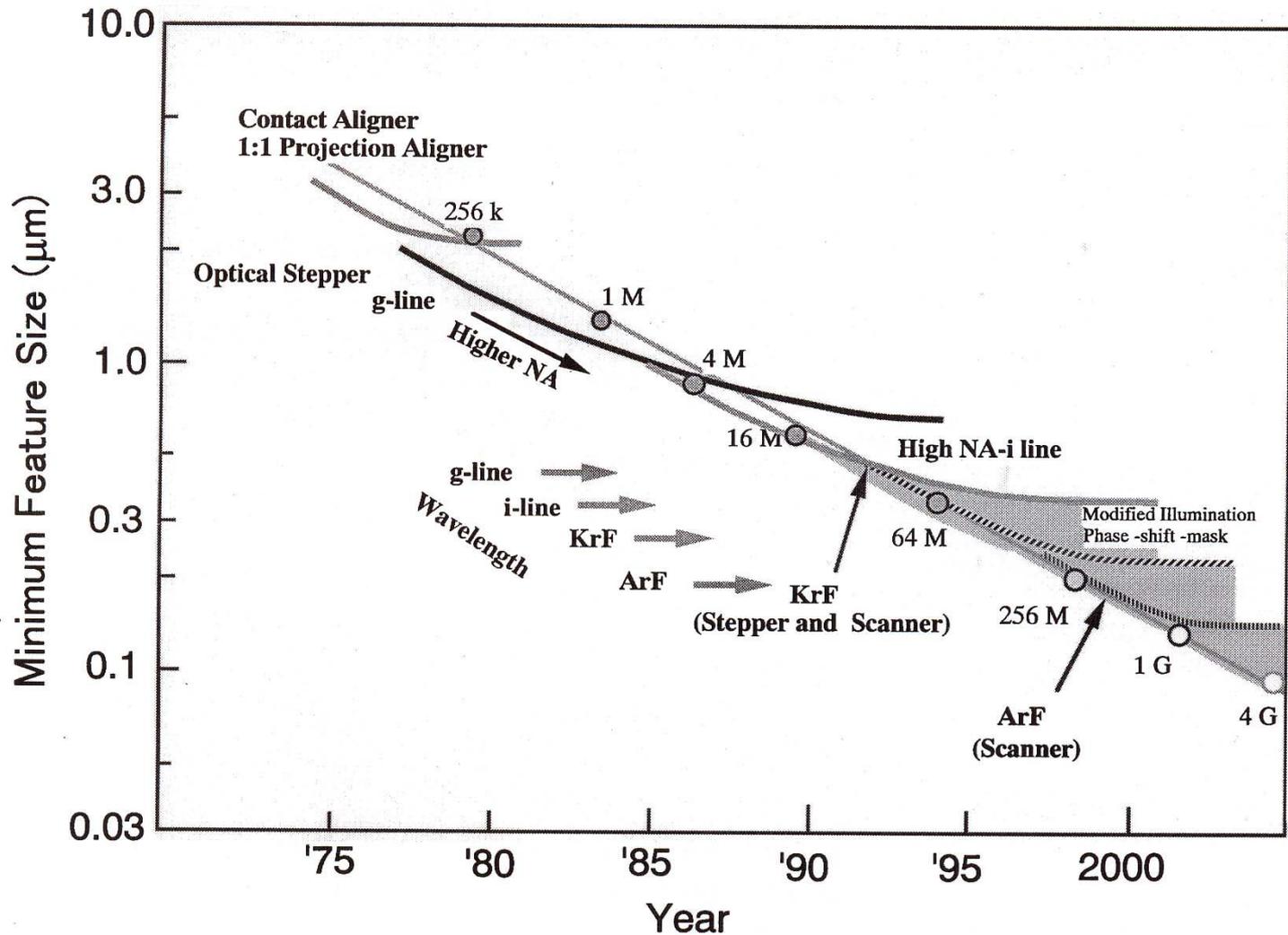
Nano-fluorescent Light



Single Ion Track



Field Emitter



Miniaturization trends of DRAM pattern size and development of optical lithographic tools. DRAM capacity is given in bits.

First Year of IC Production (year)

2004

2007

2010

2013

2016

Dram Half Pitch (nm)

65

45

32

22

193 nm
157 nm
193 nm immersion
EPL, PEL

157 nm
193 nm immersion
ML2
EPL, PEL

157 nm immersion
EUV
EPL
ML2

EUV
EPL
ML2
Imprint lithography

Optical

Post-optical

PEL: Proximity Electron Lithography
EPL: Electron Projection Lithography
EUV: Extreme Ultraviolet
ML2: Maskless Lithography

DWEB

*ITRS2003

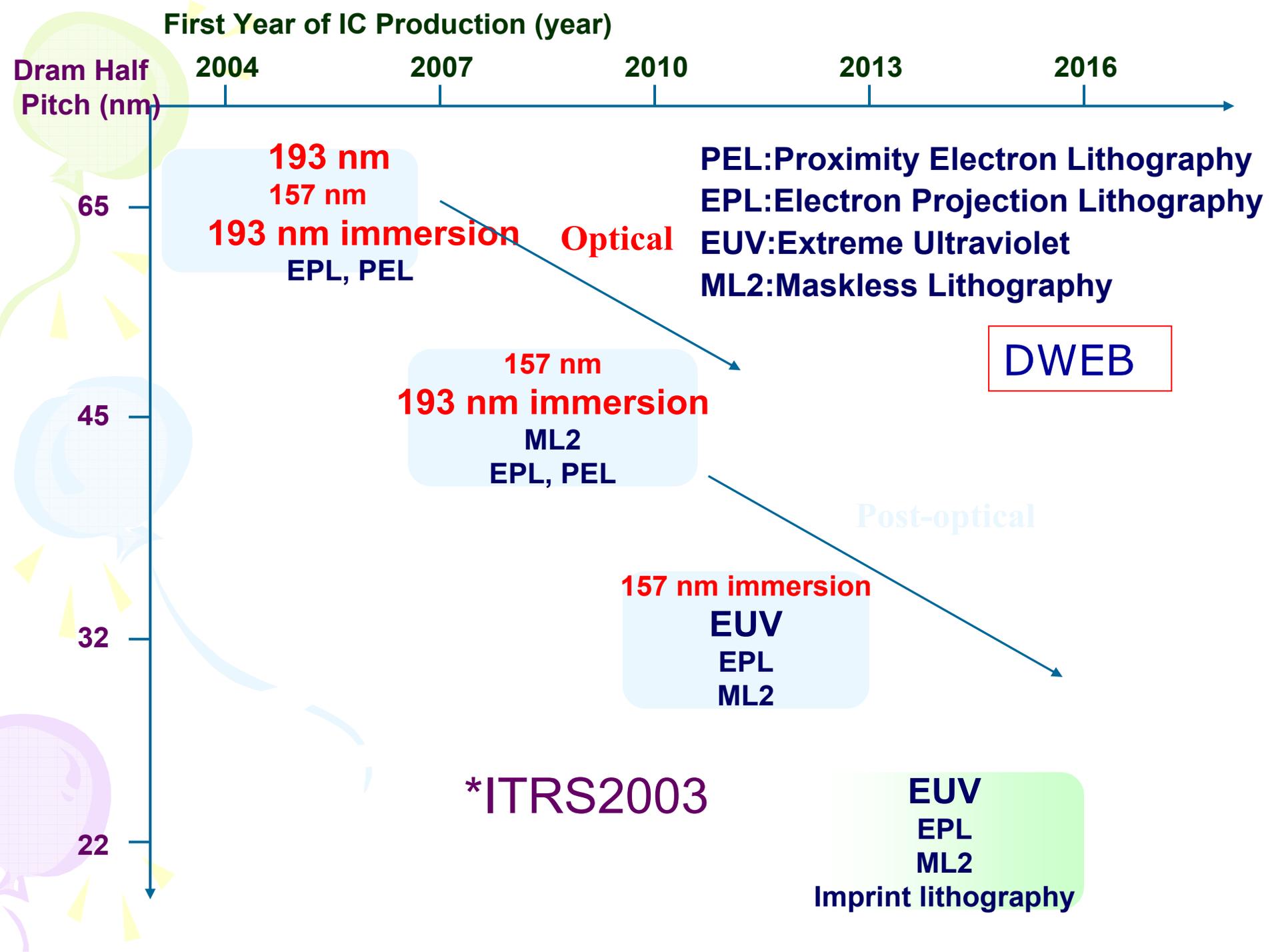


Table 78a Resist Requirements—Near-term

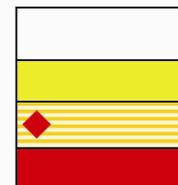
| <i>Year of Production</i> | 2003 | 2004 | 2005 | 2006 | 2007 | 2008 | 2009 |
|---|---|-----------------------|-----------------------|-----------------------|-----------------------|-----------------------|----------------------|
| <i>Technology Node</i> | | <i>hp90</i> | | | <i>hp65</i> | | |
| <i>DRAM ½ Pitch (nm)</i> | 100 | 90 | 80 | 70 | 65 | 57 | 50 |
| <i>MPU/ASIC Metal 1 (M1) ½ Pitch (nm)</i> | 120 | 107 | 95 | 85 | 76 | 67 | 60 |
| <i>MPU/ASIC ½ Pitch (nm) (un-contacted gate)</i> | 107 | 90 | 80 | 70 | 65 | 57 | 50 |
| <i>MPU Gate in resist Length (nm)</i> | 65 | 53 | 45 | 40 | 35 | 32 | 28 |
| <i>MPU Gate Length after etch (nm)</i> | 45 | 37 | 32 | 28 | 25 | 22 | 20 |
| <i>Resist Characteristics *</i> | | | | | | | |
| <i>Resist meets requirements for gate resolution and gate CD control (nm, 3 sigma) **</i> |  4.0 | 3.3 | 2.9 | 2.5 | 2.2 | 2.0 | 1.8 |
| <i>Resist thickness (nm, imaging layer) ***</i> | 250–400 | 220–360 | 200–320 | 170–250 | 160–220 | 140–200 | 130–180 |
| <i>Ultra thin resist thickness (nm)****</i> | 120–150 | 120–150 | 120–150 | 100–150 | 100–130 | 100–130 | 80-120 |
| <i>PEB temperature sensitivity (nm/C)</i> | 2.5 | 2 | 2 | 1.5 | 1.5 | 1.5 | 1.5 |
| <i>Backside particles (particles/m² at critical size, nm)</i> | 2000 @ 150 | 2000 @ 150 | 1500 @ 100 | 1500 @ 100 | 1500 @ 100 | 1500 @ 100 | 1000 @ 50 |
| <i>Defects in spin-coated resist films†</i> #/cm ² | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 |
| (size in nm) | 60 | 55 | 50 | 45 | 40 | 35 | 30 |
| <i>Defects in patterned resist films, gates, contacts, etc.</i> #/cm ² | 0.07 | 0.06 | 0.05 | 0.04 | 0.04 | 0.03 | 0.03 |
| (size in nm) | 60 | 55 | 50 | 45 | 40 | 35 | 30 |
| <i>Line Width Roughness (nm, 3 sigma) <8% of CD *****</i> |  3.6 | 3.0 | 2.6 | 2.2 | 2.0 | 1.8 | 1.6 |

Manufacturable solutions exist, and are being optimized

Manufacturable solutions are known

Interim solutions are known

Manufacturable solutions are NOT known



***ITRS2003**

What limits the resolution of EB lithography

The limit of the resolution of the high sensitive EB resists is key problem for the future lithographies (both EB and EUV lithograph).

It is also essence for the development of nanotechnology.

Effects of developer,
Development conditions

Beam size (< 1 nm)

Polymer size effects
Cluster size effects

Intermolecular forces
during development

Acid diffusion

Forward- and back-
scattered electrons

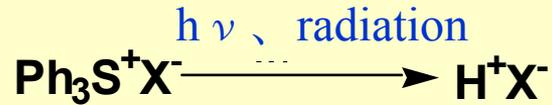
**Reaction
mechanisms**

EB can be focused less than 1 nm. However, technical barriers exist around 30-50 nm for mass production type resist pattern. **Why?**

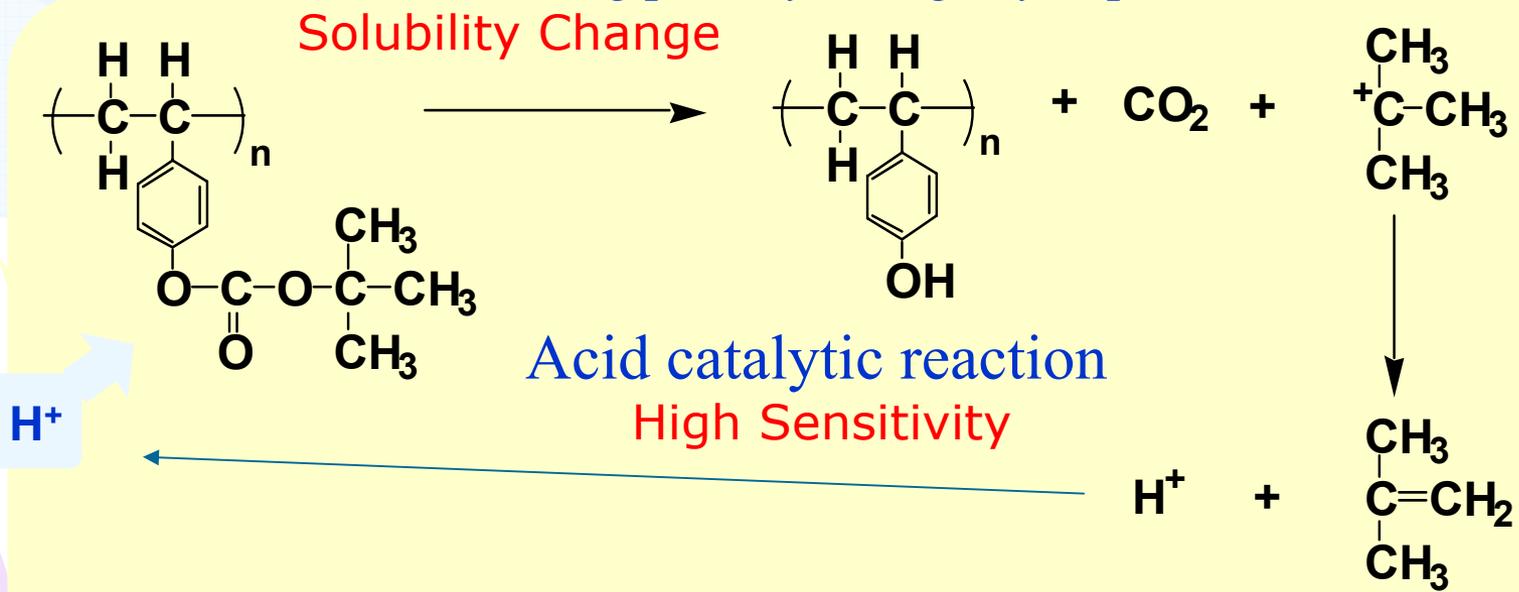
More important problem is critical dimension (CD) and Line edge roughness (LER).

Chemically amplified resist

Generation of acid by exposure



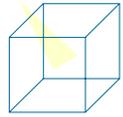
Pattern formation utilizing polarity change by deprotection reaction



× Polymerization, graft polymerization, and others

Acid generation mechanism – Ionization channel

Electron beam, X-ray, EUV



Base polymer



Acid generator



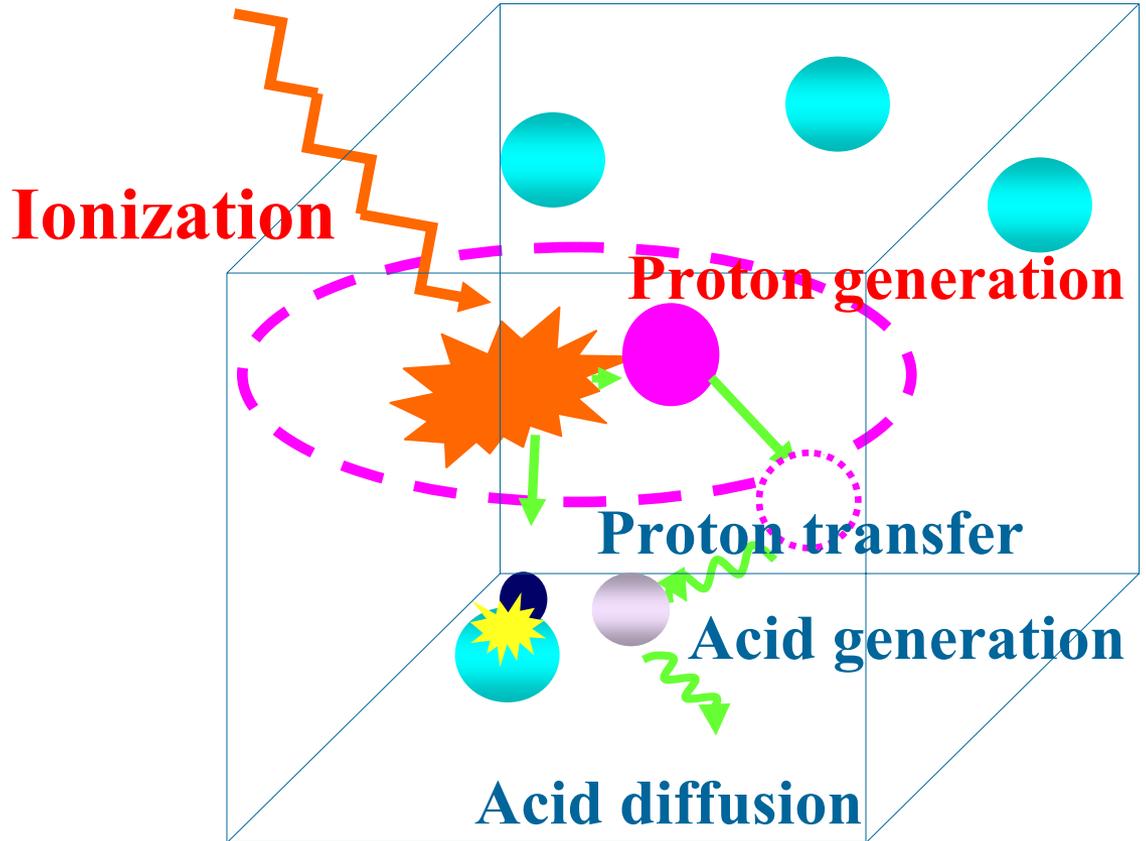
Counter anion



Proton



Electron



Reaction control in nanospace --Time space translation

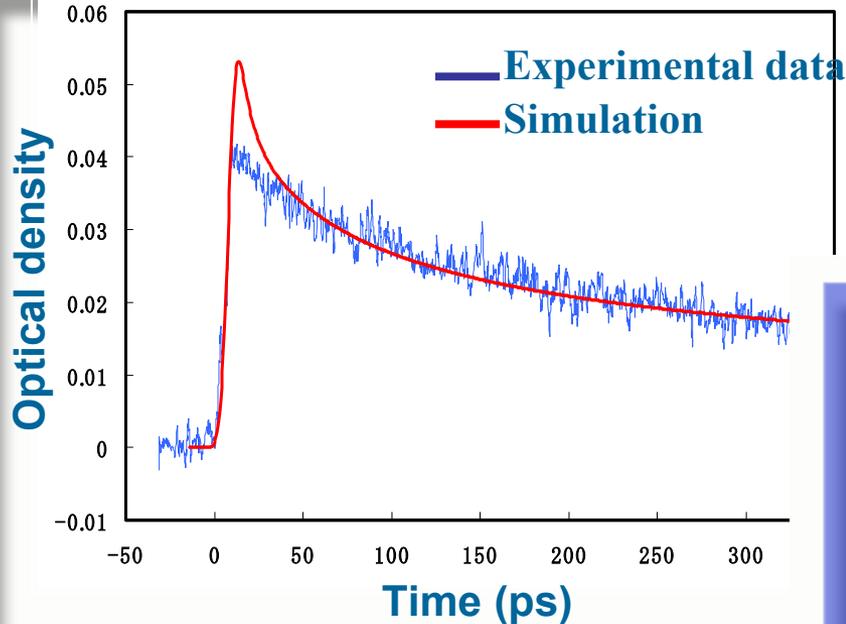
A. Saeki et al. Jpn. J. Appl. Phys. 41 (2002) 4213.

It is essential to minimize the displacement between energy deposition point and reaction point.

Time-space translation

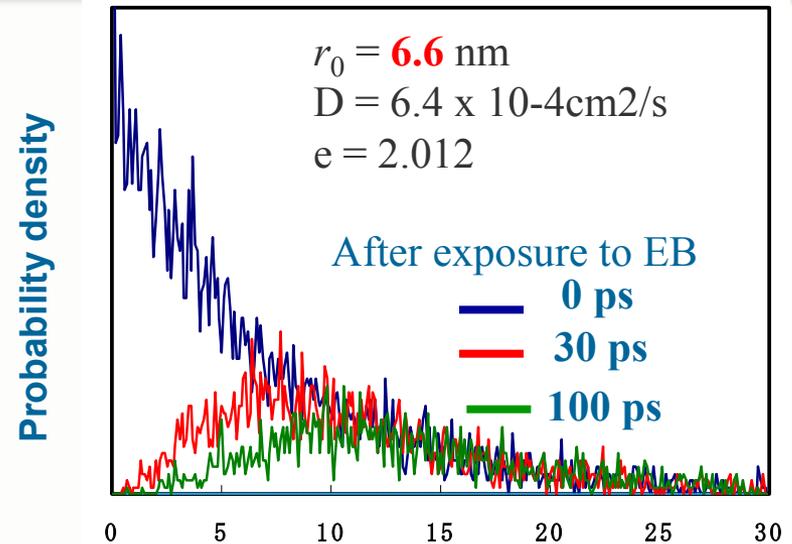
Translation of picosecond temporal data to nanometer spatial data

(So far, available only at ISIR)



Experimental data obtained in the femtosecond pulse radiolysis and simulation

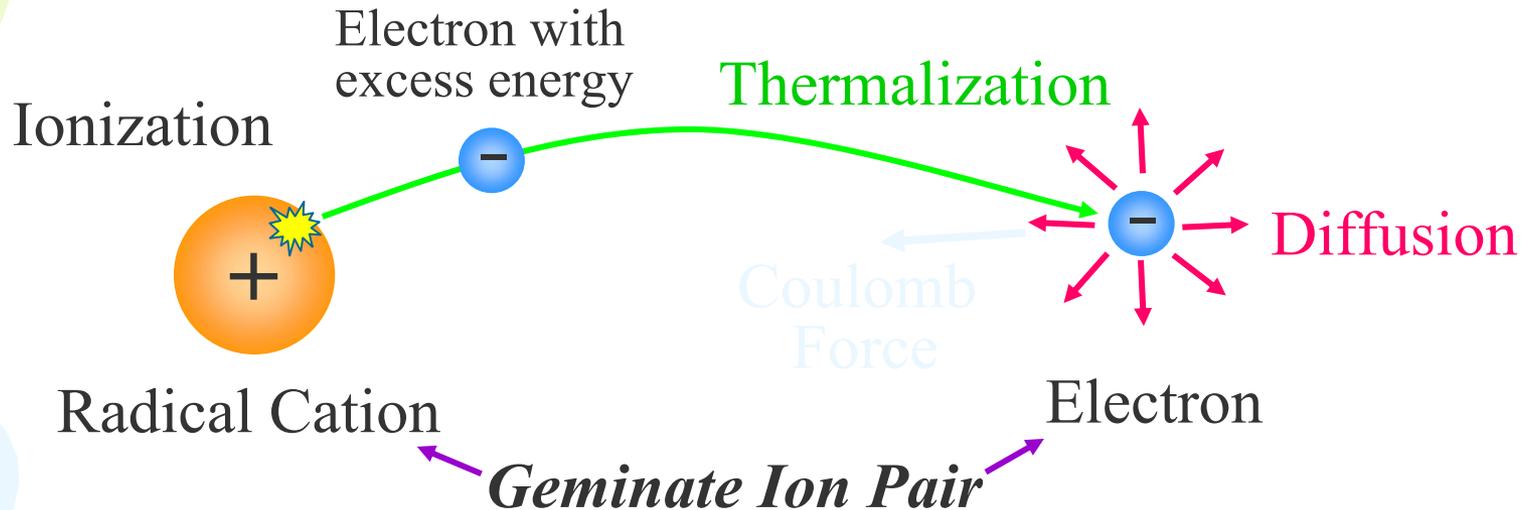
Spatial separation between electron and cation radical causes the displacement between energy deposition point and reaction point. For the nanotechnology, it is essential to decrease the displacement.



The distance between cation radical and electron [nm]

Change of distance between electron and cation radical generated by electron beam irradiation.

Electron dynamics in early processes of radiation chemistry



Smoluchowski equation

$$\frac{\partial w}{\partial t} = D \nabla \left(\nabla w + w \frac{1}{k_B T} \nabla V \right)$$

w : Probability density of electrons

k_B : Boltzmann constant

V : Coulomb potential

T : Absolute temperature

D : Sum of diffusion coefficient

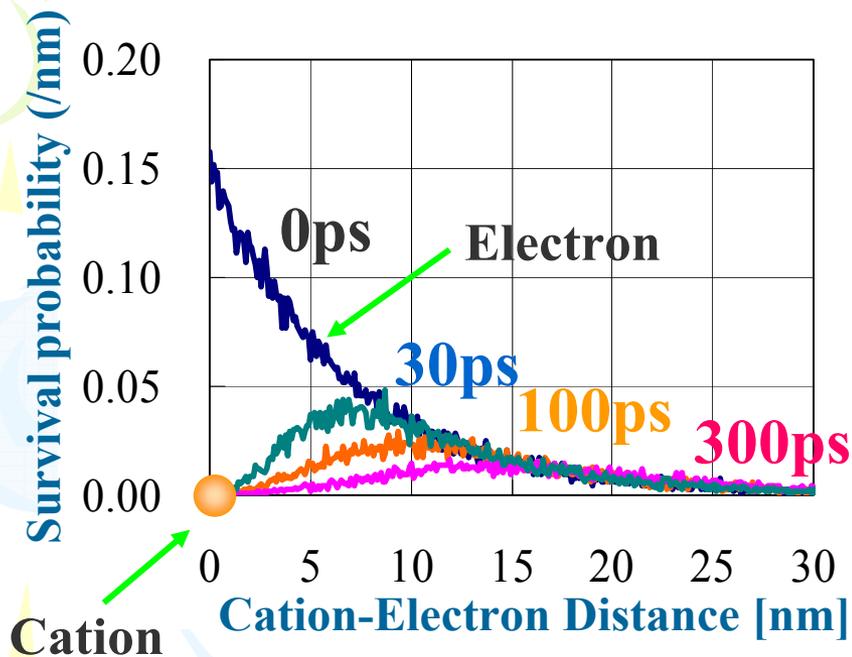
Initial distribution function

$$f(r, r_0) = \frac{1}{r_0} \exp\left(-\frac{r}{r_0}\right)$$

r : Distance between radical cation and electron

r_0 : Initial separation distance on average

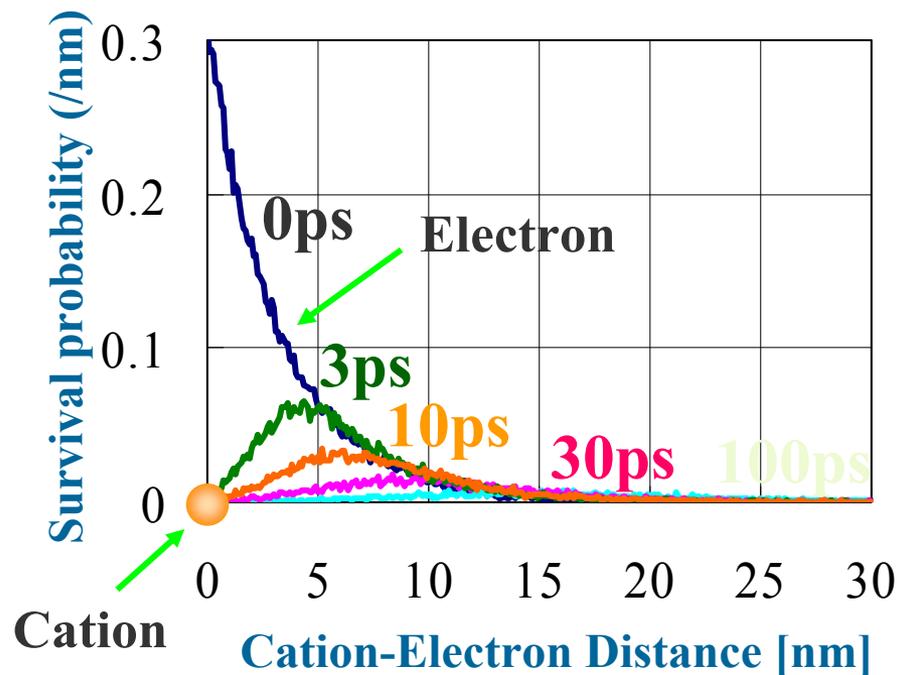
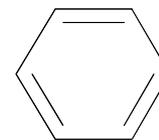
n-Dodecane



Distribution of Electron

$$r_0 = 6.6 \text{ nm}$$
$$D = 6.4 \times 10^{-4} \text{ cm}^2/\text{s}$$
$$\varepsilon = 2.012$$

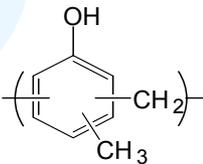
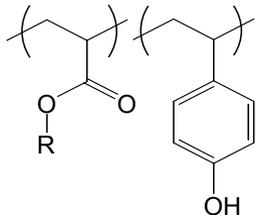
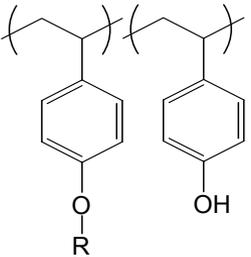
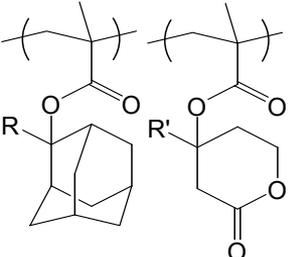
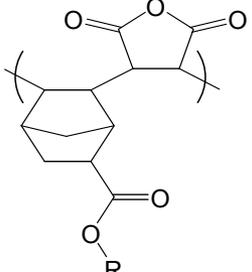
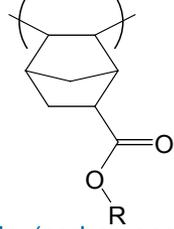
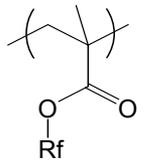
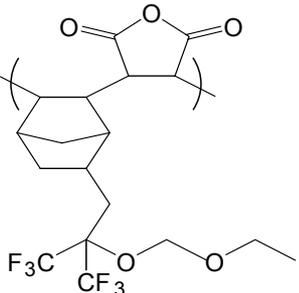
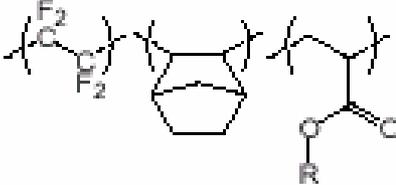
Benzene



Distribution of Electron

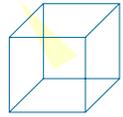
$$r_0 = 3.2 \text{ nm}$$
$$D = 2.8 \times 10^{-3} \text{ cm}^2/\text{s}$$
$$\varepsilon = 2.284$$

Base Polymers for Lithographies

| g / i | KrF | ArF | F ₂ |
|--|--|--|--|
| Aromatic ring | | Alicyclic groups Cyclic olefine polymer | F polymer |
|  <p data-bbox="251 906 540 942">Nobolak resin</p> |  <p data-bbox="579 721 869 778">Poly(acrylate -hydroxystyrene)</p>  <p data-bbox="531 1120 917 1156">Ploy(p-hydroxystyrene)</p> |  <p data-bbox="1004 664 1236 699">Ploy(methacrylate)</p>  <p data-bbox="956 978 1265 1042">Ploy(norbornene - maleic anhydride)</p>  <p data-bbox="1014 1270 1226 1299">Ploy(norbornene)</p> |  <p data-bbox="1458 592 1690 628">Ploy(methacrylate)</p>  <p data-bbox="1400 949 1719 1021">Ploy(norbornene - maleic anhydride)</p>  <p data-bbox="1468 1249 1671 1285">[TFE/NB type]</p> |

Acid generation mechanism – Ionization channel

Electron beam, X-ray, EUV



Base polymer



Acid generator



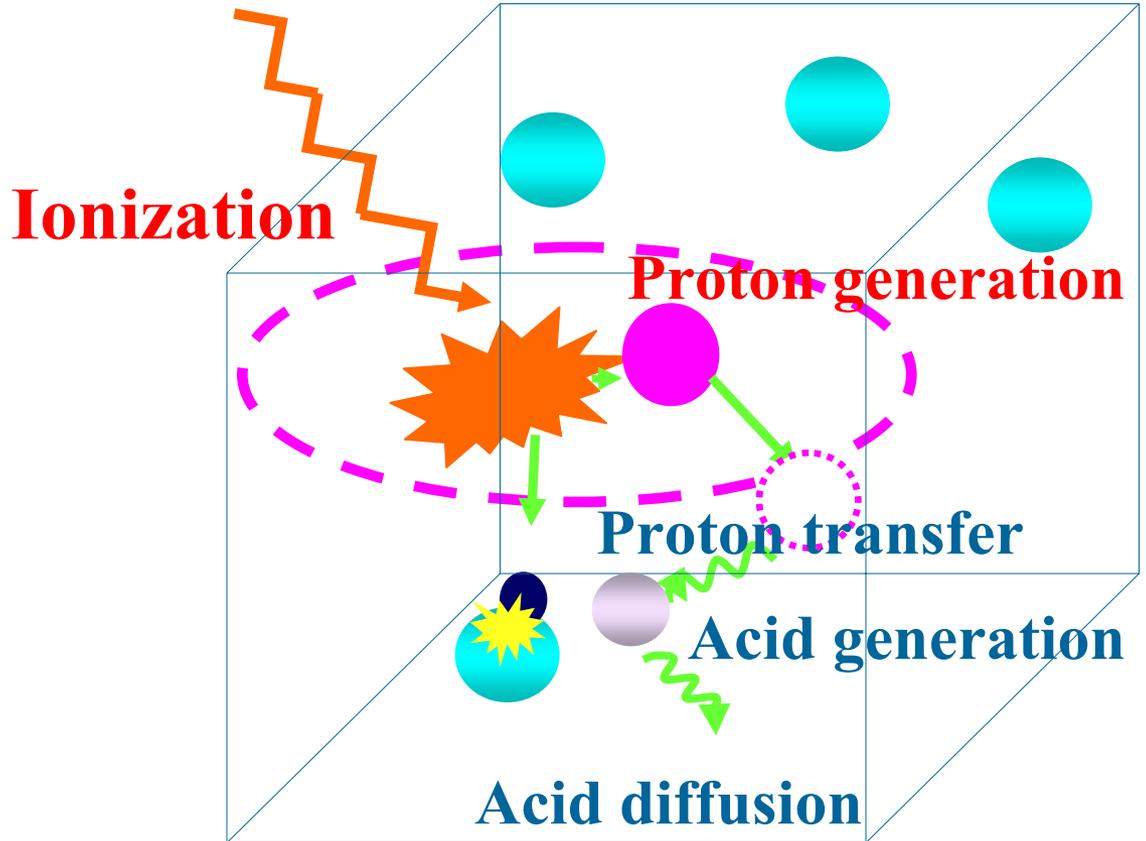
Counter anion



Proton



Electron



Formulation

The reaction of acid generators with electrons:

$$\frac{\partial w}{\partial t} = -kCw$$

Effective reaction radius of acid generators:

$$k = 4\pi RD$$

Electron dynamics in resist materials:

$$\frac{\partial w}{\partial Dt} = \nabla \left(\nabla w + w \frac{1}{k_B T} \nabla V \right) - 4\pi RCw$$

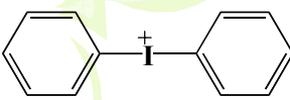
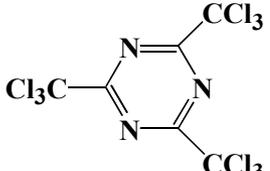
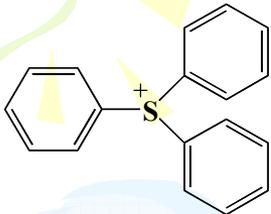
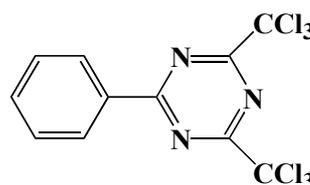
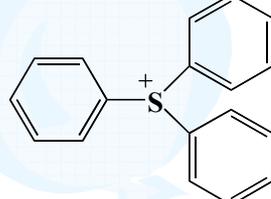
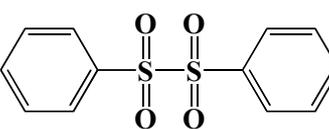
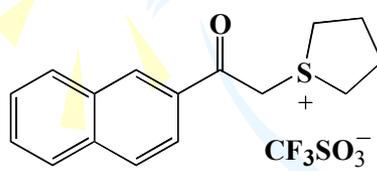
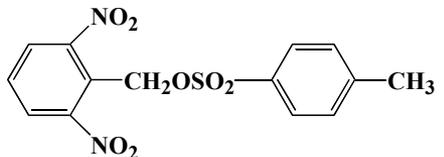
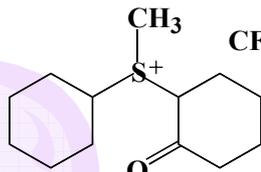
w : probability density of electrons

k : rate constant of reaction of acid generator with electrons

C : concentration of acid generators

R : effective reaction radius

Effective reaction radii of acid generators

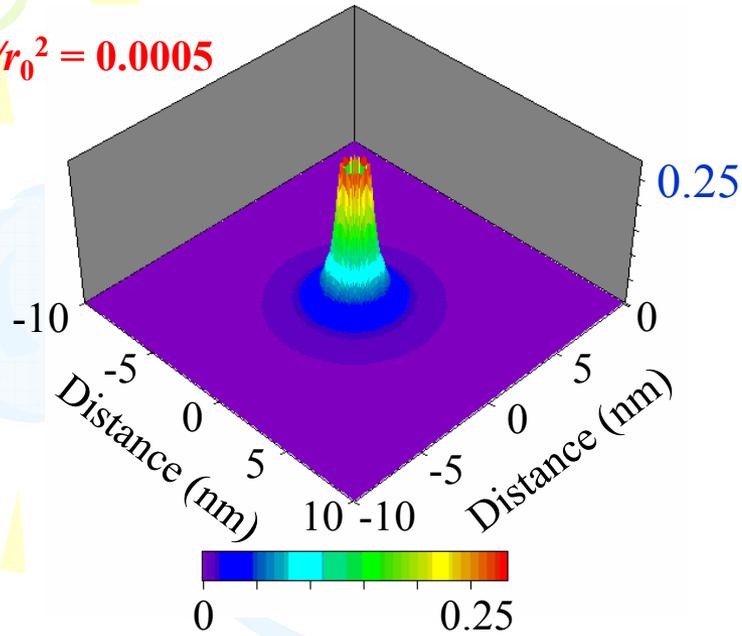
| Acid generator | k ($M^{-1}s^{-1}$) | R (nm) | Acid generator | k ($M^{-1}s^{-1}$) | R (nm) |
|---|------------------------|------------|--|------------------------|-------------|
|  <chem>CCCC[N+](CCCC)(CCCC)CCCC.[O-]S(=O)(=O)C(F)(F)F</chem> | 2.4×10^{10} | 2.1 |  <chem>ClC1=NC(Cl)=NC(Cl)=N1</chem> | 2.2×10^{10} | 1.9 |
|  <chem>CCCC[S+](CCCC)(CCCC)CCCC.[O-]S(=O)(=O)C(F)(F)F</chem> | 2.7×10^{10} | 2.4 |  <chem>ClC1=NC(Cl)=NC(Cl)=N1c2ccccc2</chem> | 2.5×10^{10} | 2.2 |
|  <chem>CCCC[S+](CCCC)(CCCC)CCCC.[O-]S(F)(F)(F)(F)(F)F</chem> | 2.5×10^{10} | 2.2 |  <chem>O=S(=O)(c1ccccc1)c2ccccc2</chem> | 1.1×10^{10} | 0.96 |
|  <chem>C1=CC=C2C=CC=C1C=C2C(=O)CS[+]1CCCC1.[O-]S(=O)(=O)C(F)(F)F</chem> | 1.9×10^{10} | 1.7 |  <chem>Cc1ccc(cc1)OS(=O)(=O)Cc2cc([N+](=O)[O-])ccc2</chem> | 2.2×10^{10} | 1.9 |
|  <chem>C1CCCCC1[S+](C)C1CCCCC1.[O-]S(=O)(=O)C(F)(F)F</chem> | 1.6×10^{10} | 1.4 | | | |

$k = 4\pi RD$

R : Effective reaction radius of acid generators
 D : Diffusion coefficient of solvated electrons in methanol ($1.5 \times 10^{-9} \text{ m}^2\text{s}^{-1}$)
 k : Rate constant of reaction of acid generators with solvated electron in methanol

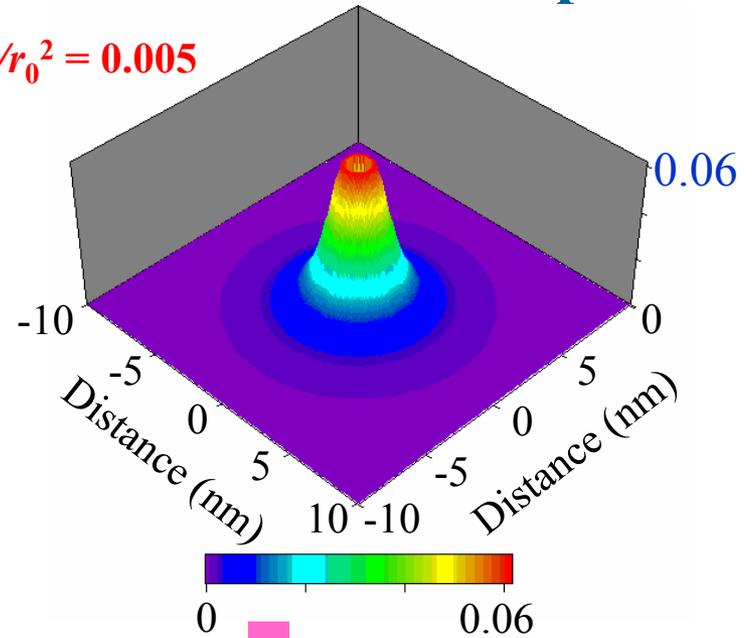
Distribution change of electrons around ionization point

$$Dt/r_0^2 = 0.0005$$



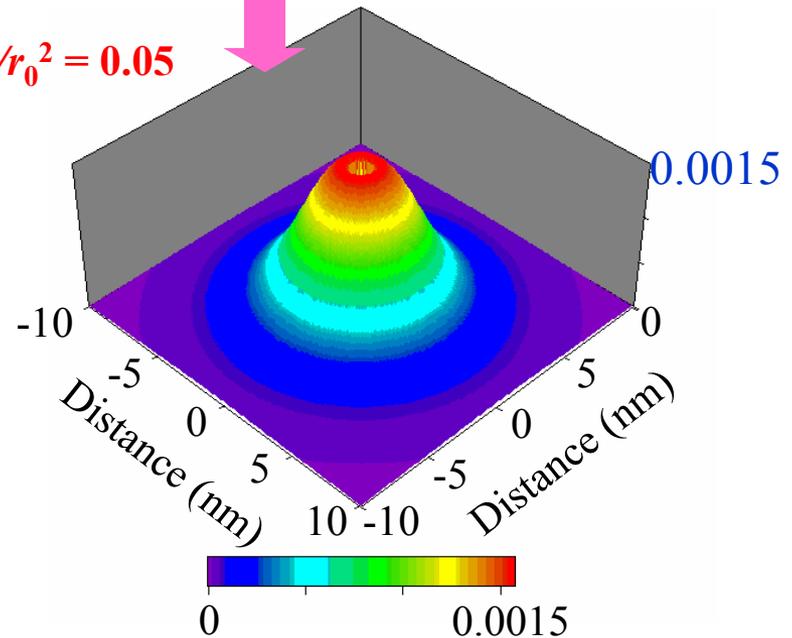
Probability density (/nm³)

$$Dt/r_0^2 = 0.005$$



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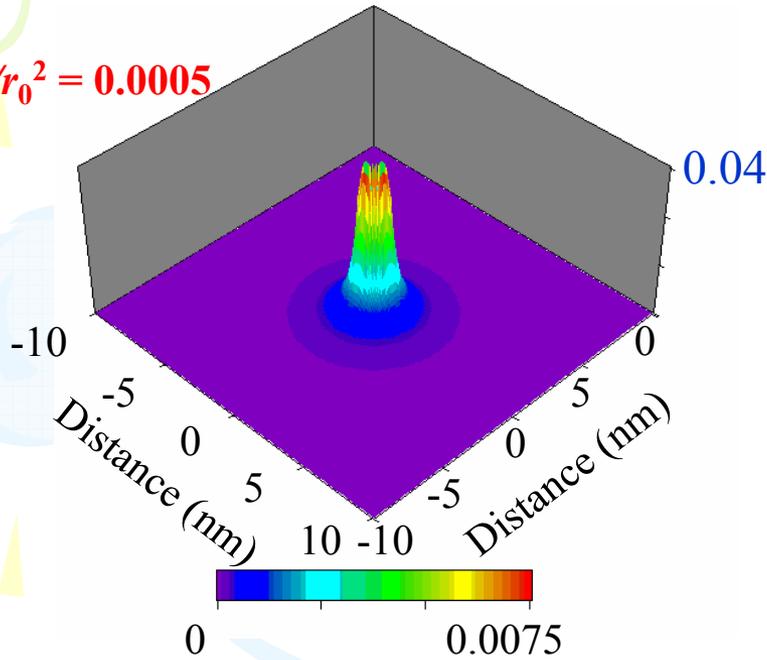


Probability density (/nm³)

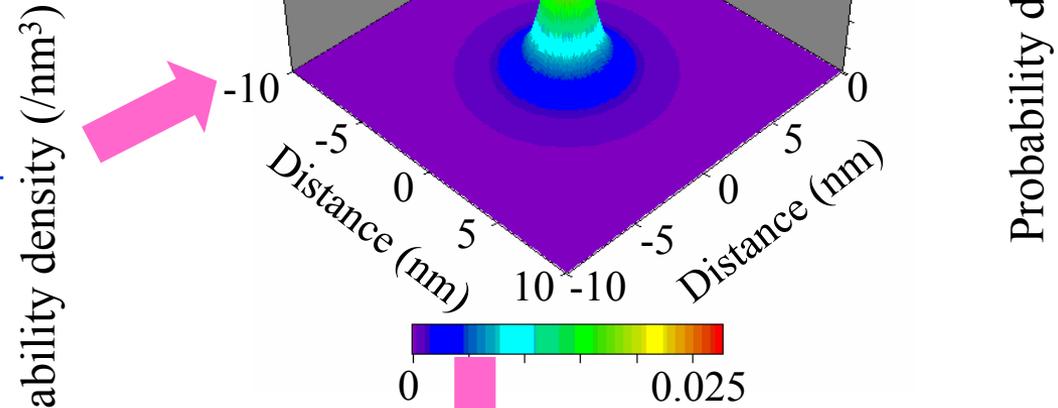
The distribution of probability density of electrons in the x-y plane. The coordination of ionization point is the origin.

Counter anion distribution around ionization point

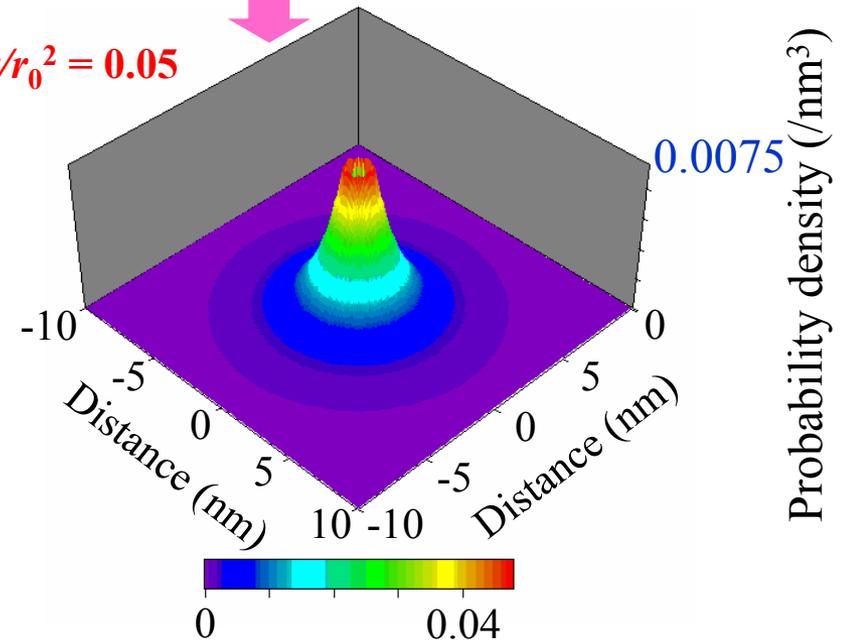
$$Dt/r_0^2 = 0.0005$$



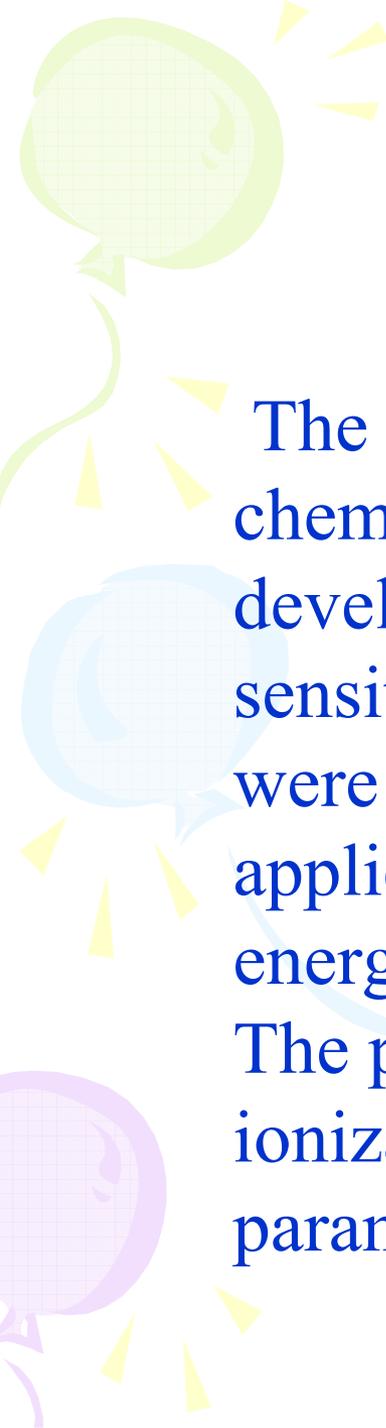
$$Dt/r_0^2 = 0.005$$



$$Dt/r_0^2 = 0.05$$



The distribution of probability density of counter anions in the x-y plane. The coordination of ionization point is the origin.



Conclusion 2

The elucidation of the reaction mechanisms of chemically amplified resists is very important in the development of the resists with at least both high sensitivity and high space resolution. Our findings were integrated to a simulation model. This model is applicable to exposure sources which have higher energy than ionization potential of resist materials. The probability density of acid distribution around ionization point was simulated with a typical parameter set.